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Page 1 of 7 Chemistry  
104 ANSWER KEY

Group Activity:  
Arrhenius Relationships  
and Reaction

Mechanisms 1. Rate  
constants at various  
temperatures for two  
reactions were  
measured and are  
graphed below:

Experimental Data

1/Temperature	0.0085				
	0.0075	0.0065	0.0055		
	0.0045	0.0035	ln(k) for		
Reaction 1	3	2.5	2	1.5	
	1	0.5	ln(k) for ...		



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perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

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**49 Balancing  
Chemical Equations  
Worksheets [with  
Answers]**

Simplifying and Solving  
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Solving Equations

Golden Rule of Algebra:

“Do unto one side of  
the equal sign as you  
will do to the other...”

Whatever you do on  
one side of the equal  
sign, you **MUST** do the  
same exact thing on  
the other side. If you  
multiply by  $-2$  on the

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left side, you have to multiply by -2 on the other. If

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**Answer**

(b) Give one or more examples of how chemistry has affected another science. View Answer Explain what chemophobia is, the "symptoms", and how you would help people who are afflicted better ...

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Answer: A balanced  
*Page 16/28*



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chemical equation has an equal number of atoms of different elements in the reactants and products. The chemical equations should be balanced to satisfy the law of conservation of mass. Question 5 Translate the following statements into chemical equations and then balance them.

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**Class 10 Science**  
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...

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To balance the chemical equation, you must add coefficients in front of the chemical formulas in the equation. You cannot add or change subscripts!!

Procedures. Determine the number of atoms listed in the equation for each element. Pick

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an element that is not equal on both sides of the equation.

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Scientists are  
investigating the

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production of energy using hydrogen-2 nuclei (deuterons) and hydrogen-3 nuclei (tritons). The balanced equation below represents one nuclear reaction between two deuterons.  $2\ ^1_1\text{H} + 2\ ^1_1\text{H} \rightarrow\ ^3_2\text{He} + 1\ ^0_0\text{n} + 5.23 \times 10^{-13}\text{ J}$ . Identify the type of nuclear reaction represented by the equation.

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**mistry\_review\_packe  
t\_key.pdf - 1 Base ...**

Chemistry-Nuclear  
Packet Worksheet #3:  
Bombardment  
Reactions Name: k L  
Hour: Page 6 So far,  
the equations we have  
written have involved  
natural radioactive  
decay and therefore  
natural transmutation  
(changing of one  
element into another  
element). However, we  
have learned to cause  
transmutation by



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bombardment of nuclei  
with high-energy  
particles.

**I,**

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chemical equations for  
what happens when a)

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H<sub>3</sub>O<sup>+</sup> and b) OH<sup>-</sup> are  
added to a HNO<sub>3</sub>,  
/NaNO<sub>3</sub>, buffer. a. b.

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