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A solution with a pH value of 12 is a _____. Base. Atoms are composed of _____.

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The correct answer is (B). The hydrogen

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bonds between water molecules are able to form, break and re-form frequently. Water's life-sustaining properties are a result of hydrogen bonding that orders molecules into a higher level of structural organization.

AP Biology Practice Test: Unit 1 — Chemistry of Life ...

1. Compare protons, electrons, and neutrons with respect to charge, location, and mass. 2. What relationship exists between the mass number of an element and isotopes of that element? 3. Explain the difference between ionic and covalent bonds. 4. Compare and contrast adhesion and cohesion.

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Catalase helps prevent a toxic buildup of hydrogen peroxide in cells by breaking it down into water and oxygen gas.

Several students conduct an experiment to test the effects of pH on the activity of catalase. Each test tube contains a solution of hydrogen peroxide and water at various pH levels. The liver tissue is a source of catalase.

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Chapter 2: The Chemistry of Life

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Chapter 2: The Chemistry of Life Biology: McDougal Littel pages 34-64. Below you find the classroom assignments and PPT's used for Chapter 2, The Chemistry of Life. You may use this website for access to PPT's, guided notes, and make up assignments. The Chemistry of Life Assignments.

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Chapter 2: Introduction to the Chemistry of Life Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things.

Chapter 2: Introduction to the Chemistry of Life ...

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